# organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 200 KMean  $\sigma$ (C–C) = 0.005 Å Disorder in main residue R factor = 0.035 wR factor = 0.131 Data-to-parameter ratio = 13.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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Sperminium bis(tri-*tert*-butoxysilanethiolate)-2-propanol (1/2)

The title compound,  $C_{10}H_{28}N_4^{2+}\cdot 2C_{12}H_{27}O_3SSi^-\cdot 2C_3H_8O$ , has been obtained by the reaction of (<sup>*t*</sup>BuO)<sub>3</sub>SiSH with spermine in the presence of 2-propanol. The sperminium cation sits on an inversion centre at the mid-point of the central C–C bond. The three species are linked together in infinite chains, mainly by N–H···S hydrogen bonds.

### Comment

Until now, spermine, despite its biological importance, has not been structurally characterized by X-ray diffraction methods. Only three structures containing the spermine residue can be found in the Cambridge Structural Database (Version 5.24, updates 1–3; Allen, 2002). These are complex salts with copper(II) perchlorate (Boggs & Donohue, 1975), copper(II) sulfate (Maluszynska *et al.*, 1995) and palladium(II) chloride (Codina *et al.*, 1999).

Our scientific interest has focused recently on  $N-H\cdots S$  hydrogen-bond networks. Spermine, being a polyamine, provides a unique opportunity to explore the cooperation of several amino groups in the formation of the overall structure, including self-organization of packing in the crystal lattice. Here, we describe the molecular and crystal structure of the title compound, (I), containing a doubly protonated sperminium salt with a bulky *tert*-butoxysilanethiolate anion and two solvent molecules of 2-propanol.



2 (¹BuO)<sub>3</sub>SiS<sup>⊝</sup>, 2 <sup>i</sup>PrOH (I)

The asymmetric unit of (I) consists of one tri-*tert*-butoxysilanethiolate anion, one-half sperminium cation and one 2-propanol molecule. The sperminium cation sits on an inversion centre at the mid-point of the central C-C bond, at Wyckoff position  $h(\frac{1}{2},\frac{1}{2},\frac{1}{2})$ . The ions are linked in infinite chains. The two terminal ammonium groups of the sperminium cation form N-H···S hydrogen bonds, each -NH<sub>3</sub><sup>+</sup> group with one thiolate anion. This leads to the graph-set motif  $R_4^2(8)$  (Etter, 1990), the centre of which lies on the inversion centre at  $(0,0,\frac{1}{2})$ , *i.e.* Wyckoff position *b*.

The molecule of 2-propanol in (I) accepts one H atom from a non-terminal N atom, forming an  $N-H\cdots O$  hydrogen bond. Simultaneously, its OH group behaves as a donor to form an  $O-H\cdots S$  hydrogen bond. The S atom forms an  $S\cdots H-N$ bond with the terminal  $-NH_3^+$  group. Additionally, each Received 10 November 2003 Accepted 20 November 2003 Online 29 November 2003

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#### Figure 1

The molecular structure of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. All H atoms bonded to C atoms have been omitted for clarity. The asymmetric unit is shown, together with the other half of the cation. Only one component of the disordered tert-butyl group is included

terminal ammonium group forms an intramolecular N- $H \cdots N$  hydrogen bond with one non-terminal amine group. These interactions may be described as a set of first-level graphs C(3), C(3), C(3), S(6), or, taken together, as the graphset motif  $R_4^3(8)$  [adjacent to the above-mentioned N-H···S  $R_4^2(8)$  motif].

The S···N distances in (I) lie in the range 3.262 (3)-3.265 (3) Å, comparable with values observed in dicyclohexylammonium benzenethiolate (Baranowska et al., 2003), as well as in zinc and cobalt benzenethiolate complexes (Chung et al., 1991a,b).

# **Experimental**

All synthetic procedures were carried out under argon, using standard Schlenk techniques. Tri-tert-butoxysilanethiol (Piekos & Wojnowski, 1962) was prepared according to literature procedures. Solvents were dried by standard methods and distilled under argon prior to use. ('BuO)<sub>3</sub>SiSH (3 mmol) was added to a solution of spermine (120 mg, 0.6 mmol) [NH<sub>2</sub>(CH<sub>2</sub>)<sub>3</sub>NH(CH<sub>2</sub>)<sub>4</sub>NH(CH<sub>2</sub>)<sub>3</sub>NH<sub>2</sub>] in toluene (10 ml). On allowing the solution to stand for ca one week at 273-278 K, small transparent crystals were formed. These were collected and recrystallized from 2-propanol, giving small crystals of (I) of good quality, with a sharp m.p. of 357-358 K. Analysis found: C 53.6, H 10.9, N 7.2, S 8.3%; calculated for C<sub>17</sub>H<sub>41</sub>N<sub>2</sub>O<sub>3</sub>SSi: C 53.5,





H 10.8, N 7.3, S 8.4%. This result indicates loss of 2-propanol during drying

Z = 1
$D_x = 1.064 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
Cell parameters from 6
reflections
$\theta = 5.9 - 31.9^{\circ}$
$\mu = 0.19 \text{ mm}^{-1}$
T = 200 (2)  K
Block, colourless
$0.40 \times 0.30 \times 0.20$ mm

#### Data collection

Kuma KM-4 diffractometer  $\omega/2\theta$  scans Absorption correction: none 4475 measured reflections 4254 independent reflections 2954 reflections with  $I > 2\sigma(I)$  $R_{\rm int} = 0.020$ 

#### Refinement

S

42

31

Refinement on $F^2$	H atoms treated by a mixture of
$R[F^2 > 2\sigma(F^2)] = 0.035$	independent and constrained
$wR(F^2) = 0.131$	refinement
S = 1.00	$w = 1/[\sigma^2(F_o^2) + (0.0913P)^2]$
4254 reflections	where $P = (F_o^2 + 2F_c^2)/3$
310 parameters	$(\Delta/\sigma)_{\rm max} < 0.001$
-	$\Delta \rho_{\rm max} = 0.31 \ {\rm e} \ {\rm \AA}^{-3}$

# $\Delta \rho_{\rm min} = -0.25 \text{ e } \text{\AA}^{-3}$

 $\theta_{\rm max} = 25.0^{\circ}$ 

 $h = -10 \rightarrow 10$  $k = 0 \rightarrow 14$ 

 $l = -15 \rightarrow 15$ 

3 standard reflections

every 200 reflections

intensity decay: 0.8%

### Table 1

Selected geometric parameters (Å, °).

S1-Si1	2.0657 (12)	N1-C40	1.479 (3)
Si1-O1	1.6302 (18)	C42-N2	1.472 (4)
Si1-O3	1.6312 (19)	N2-C43	1.469 (3)
Si1-O2	1.6418 (17)		
O1-Si1-S1	115.78 (7)	N2-C42-C41	111.5 (2)
O3-Si1-S1	114.98 (8)	C43-N2-C42	110.6 (2)
O2-Si1-S1	105.90 (7)	N2-C43-C44	113.5 (2)
N1-C40-C41	112.0 (2)	C43-C44-C44 <sup>i</sup>	114.7 (3)
C40-C41-C42	114.5 (2)		

Symmetry code: (i) 1 - x, 1 - y, 1 - z.

Та	ble	2							
Hydrogen-bonding geometry (Å, $^{\circ}$ ).									
D	TT	4		D	TT		TT	4	 

$D - \mathbf{H} \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N1-H1A\cdots S1^{ii}$	0.87 (3)	2.41 (3)	3.265 (3)	167 (3)
$N1 - H1B \cdot \cdot \cdot N2$	0.93 (3)	2.09 (3)	2.887 (3)	144 (3)
$N1 - H1C \cdot \cdot \cdot S1$	0.98 (3)	2.30 (3)	3.262 (3)	169 (3)
$N2-H2A\cdots O4^{iii}$	0.87 (3)	2.46 (3)	3.239 (3)	150 (3)
$O4-H4A\cdots S1^{iv}$	0.84 (3)	2.48 (3)	3.301 (2)	167 (3)

Symmetry codes: (ii) -x, -y, 1 - z; (iii) x - 1, y, z; (iv) 1 - x, -y, 1 - z.

The data are 88% complete. Measurements were not complete because, after 4–5 d, there were technical problems with icing. Reflections in the high- $\theta$  region were of poor quality. H atoms attached to N and O atoms were refined freely, whereas those attached to C atoms were refined with a riding model. C–H distances were set to 0.98 (methyl H), 0.99 (methylene H) and 1.00 Å (methine H).  $U_{\rm iso}$ (H) values were constrained to be 1.2 (1.5 for methyl and methine H) times  $U_{\rm eq}$  of the carrier atom. Atoms C21–C23 of one *tert*-butyl group are disordered over two positions with site occupancy factors of 0.628 (9) and 0.372 (9).

Data collection: *KM*-4 *Software* (Kuma Diffraction, 1996); cell refinement: *KM*-4 *Software*; data reduction: *DATAPROC* (Gałdecki *et al.*, 1996); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97

(Sheldrick, 1997); molecular graphics: *RESVIEW* (Schwenk, 1998); software used to prepare material for publication: *SHELXL*97.

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